

CBE 34487  
7/9/20

## OPTIMIZATION

- BY SOME PROCEDURE  
FOR SOME "OBJECTIVE"

WE ARE LOOKING FOR BEST  
ANSWER TO A PROBLEM

## SIMPLE CALCULUS EXAMPLE

CYLINDRICAL TANK

FOR A FIXED  $V$ , WHAT

RATIO OF  $D/H$  USES LEAST

AMOUNT OF MATERIAL

$$V = \frac{\pi D^2}{4} H = H = \frac{4V}{\pi D^2}$$

$$A = 2 \frac{\pi D^2}{4} + \pi D H$$

$$A = \frac{\pi D^2}{2} + \frac{4V}{D}$$

FIND EXTREMUM  $\frac{dA}{dD} = \pi D - \frac{4V}{D^2} = 0$

$$\frac{d^2A}{dD^2} = \pi + \frac{8V}{D^3}$$

$$D^3 = \frac{4V}{\pi}$$

SO MINIMUM

$$D = \sqrt[3]{\frac{4V}{\pi}}, \quad H = \sqrt[3]{\frac{4V}{\pi}}$$

$$\boxed{D/H = 1}$$

IF WE CAN GET THE OBJECTIVE  
FUNCTION, PRESUMABLY WE CAN  
FIND AN OR MULTIPLE EXTREMA  
BY ANALYTICAL OR NUMERICAL  
MEANS.

FOR A GIVEN FLOW REQUIREMENT,  
WHAT DIAMETER OF THE PIPE  
IS OPTIMAL

LARGER  $D \Rightarrow$

PUMPING COSTS  
GO DOWN

COST OF PIPE  
GOES UP

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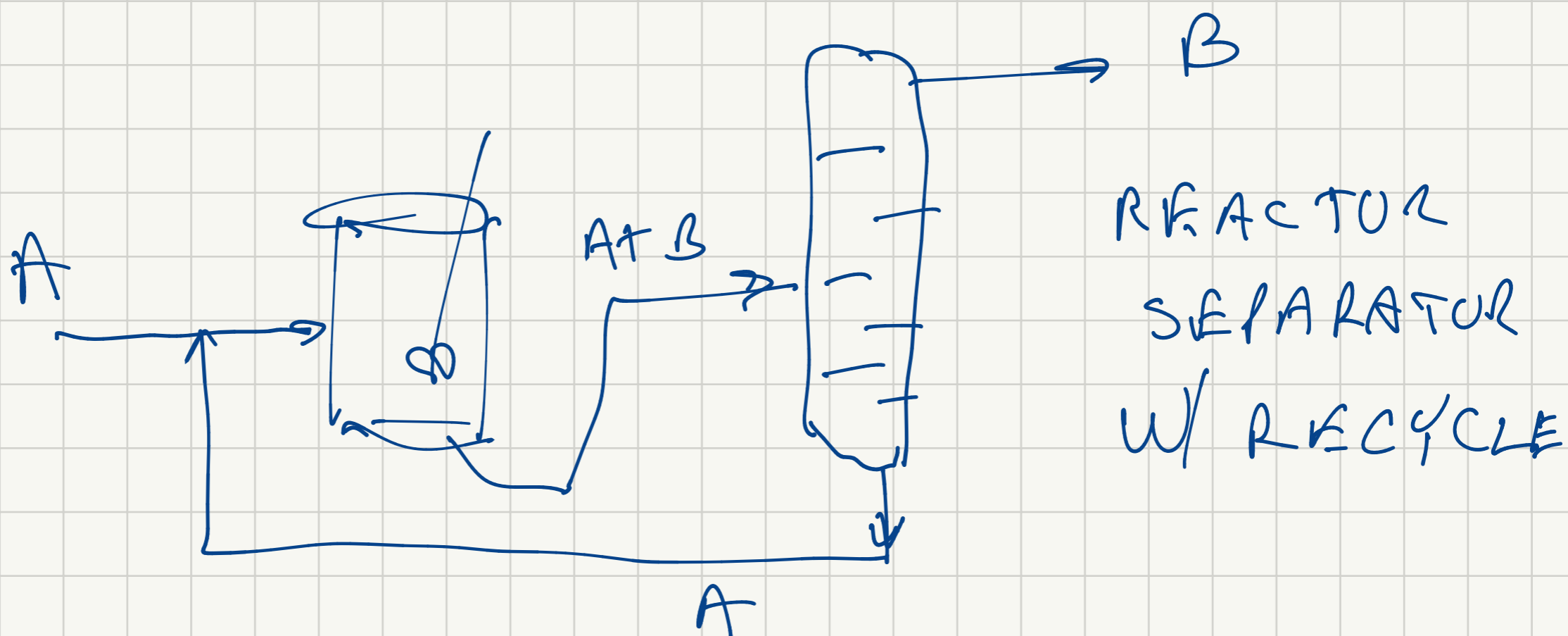
### "SIMPLE" EXAMPLE

WE WISH TO CONVERT A TO B

AND REALIZE WE NEED A

REACTION PROCESS AND

A PURIFICATION PROCESS



WE HAVE THE CONSTRAINT  
THAT WE MUST PRODUCE  
 $\beta$  MOLES/TIME

WE WANT TO DO THIS  
FOR THE MINIMUM  
POSSIBLE COST

$$\begin{aligned} \text{TOTAL COST} &= \text{COST OF EQUIPMENT} + \text{COST TO RUN PROCESS} \\ &= \text{'CAPITAL COSTS'} + \text{'OPERATING COSTS'} \end{aligned}$$

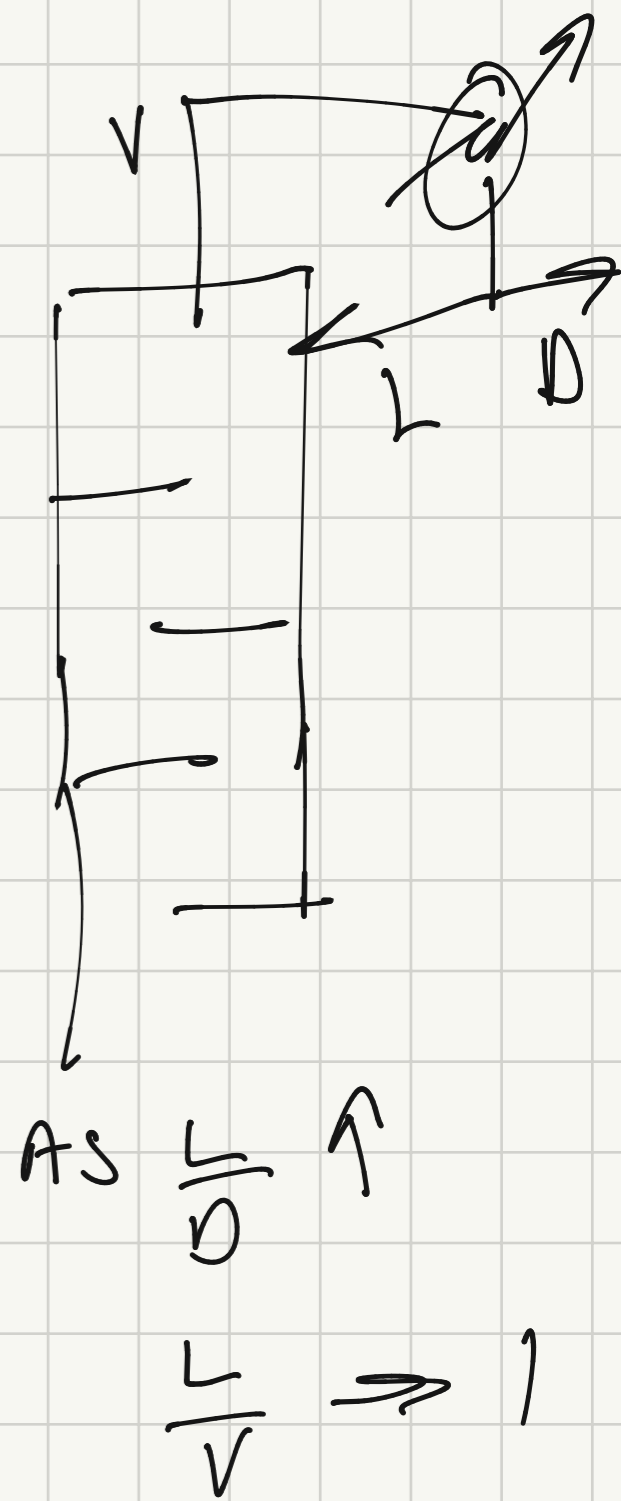
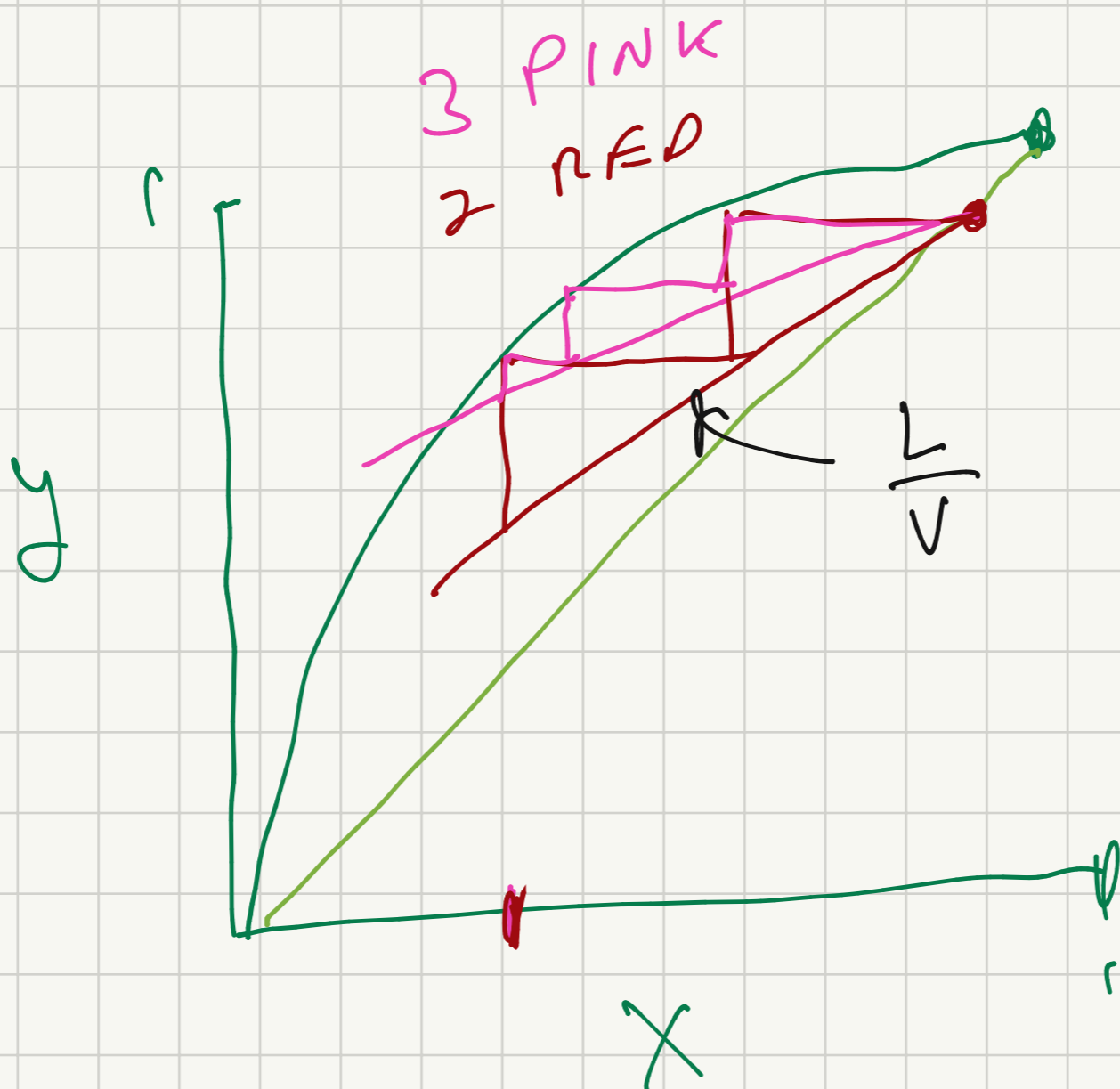
WE WISH TO CONSTRUCT A  
FUNCTION THAT INCLUDES  
THESE, THEN FIND  
MINIMUM !!

PRE SUMABLY THERE IS  
A TRADEOFF BETWEEN  
SIZE OF REACTOR & AMOUNT  
OF A THAT IS RECYCLED  
AS  $v \uparrow$  CONVERSION IS  
INCREASED SO IN THE  
LIMIT, PROCESS WOULD  
NOT HAVE RECYCLE, NOR  
NEED A SEPARATION STEP

THE LIMIT IS IMPRACTICAL  
BUT A LARGE REACTOR  
SHOULD REDUCE THE ENERGY  
NEEDED FOR SEPARATION

FOR SEPARATION, A TRADE  
OFF EXISTS BETWEEN  
ENERGY INPUT  $\downarrow$  NUMBER  
OF TRAYS NEEDED FOR  
SEPARATION

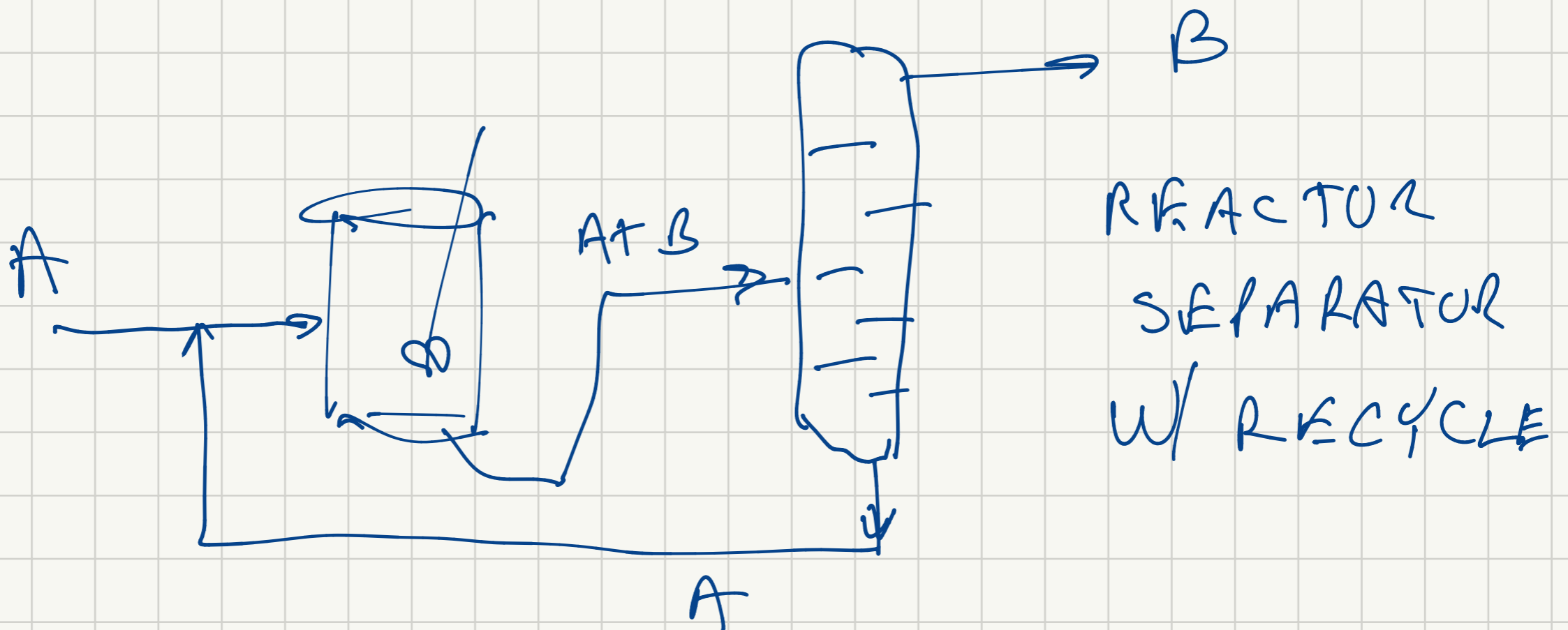
MORE  $\dot{Q}$ ,  
HIGHER  
INTERNAL  
FLOW RATES  $\Rightarrow$  FEWER TRAYS



$\downarrow$  IS FIXED SO, IF MOREL IS NEEDED  
 $V \uparrow$   $AS \uparrow$   $Q \uparrow$ .

WE WOULD LIKE INCLUDE  
THESE EFFECTS IN  
A "STRAIGHT FORWARD"  
CALCULATION

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$A \Rightarrow B$ , FIRST ORDER REACTION

FOR VAPOR-LIQUID EQUILIBRIUM,  
"CONSTANT RELATIVE  
VOLATILITY"



$$x_i \gamma_i P_i^{\text{VAP}} = y_i P$$

SUPPOSE:  $\gamma_i = 1 \rightarrow$  CHEMICALLY SIMILAR MOLECULES

$$y_i = \frac{x_i P_i^{\text{VAP}}}{P}$$

$$Q \equiv \frac{y_1/x_1}{y_2/x_2} = \text{CONSTANT}$$

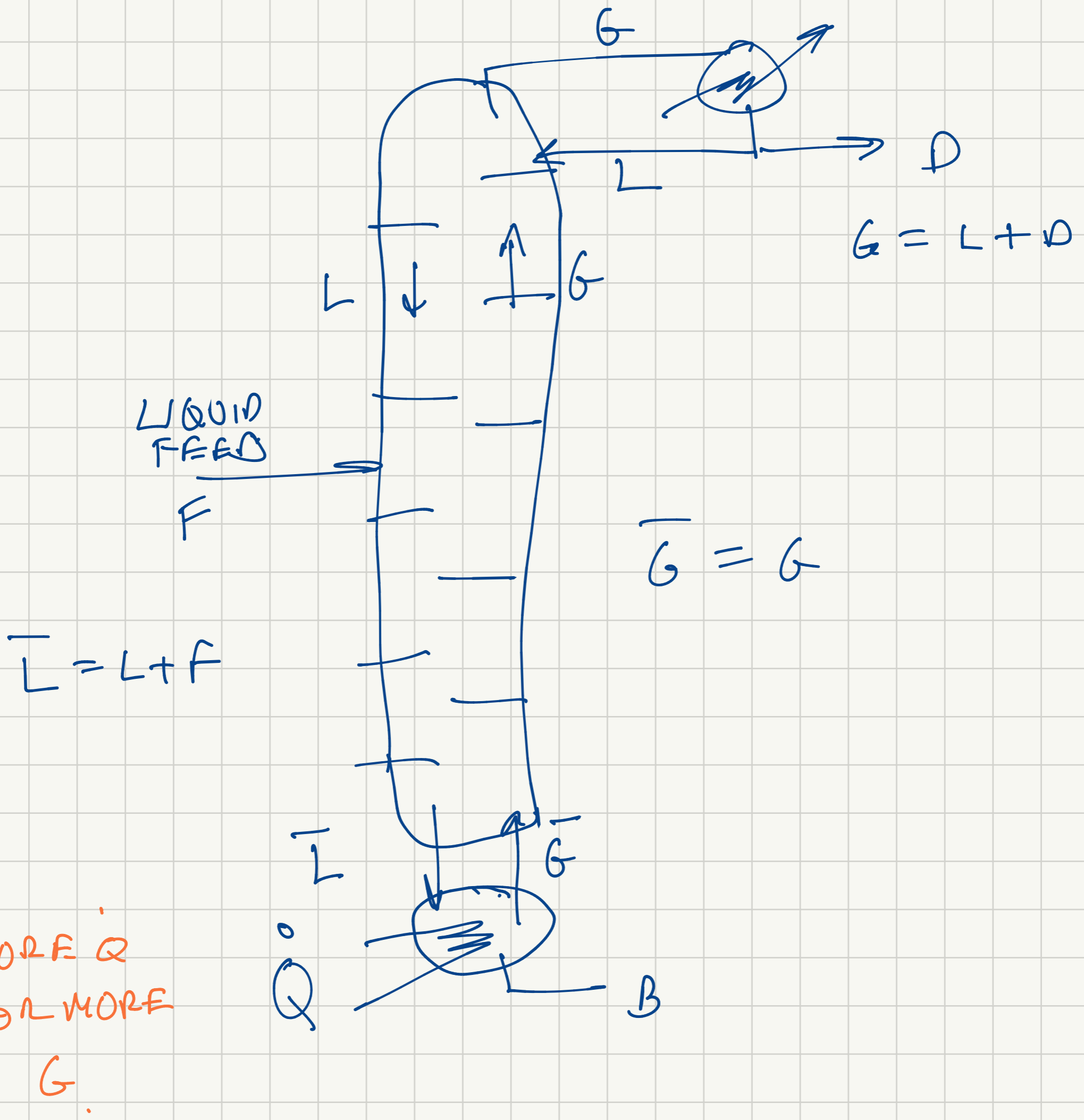
$$= \frac{P_1^{\text{VAP}}/P}{P_2^{\text{VAP}}/P} = \frac{P_1^{\text{VAP}}}{P_2^{\text{VAP}}}$$

MAY BE REASONABLE IF  $\Delta H^{\text{VAP}}$  IS ABOUT THE SAME FOR EACH COMPONENT

$$\frac{dP^{\text{SAT}}}{dT} = \frac{\Delta H^{\text{VAP}}}{T(V^{\text{V}} - V^{\text{L}})}$$

'CLAPPEYRON' EQ

# COLUMN OPERATION



MORE Q  
FOR MORE  
G.

